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Any polyatomic ion with the suffix $-ate$ uses the suffix $-ic$ as an acid. So, HNO_3 will be nitric acid. When you have a polyatomic ion with one more oxygen than the $-ate$ ion, then your acid will have the prefix $per-$ and the suffix $-ic$. For example, the chlorate ion is ClO_3^- . Therefore, $HClO_4$ is called perchloric acid.

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