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Chapter 6 Chemistry Review Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank. ____ 1. The electrons involved in the formation of a chemical bond are called a. dipoles. c. Lewis electrons. b. s electrons. d. valence electrons. ____ 2.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

6 Chemical Bonding

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molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a. molar mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg SO_3 1 g 1000 mg 1 mol $80.07 \text{ g} = 6.14 \cdot 10^{-4} \text{ mol}$ SO_3 3 b. molar mass of $PbO_2 = 239.2 \text{ g}$ $7.44 \cdot 10^4 \text{ kg}$ PbO 2 1000 g 1 kg 1 mol $239.2 \text{ g} = 3.11 \cdot 10^5 \text{ mol}$ PbO 2 c. molar mass of $CHCl_3 = 119.37 \text{ g}$ 59.1 g $CHCl_3$ 1 mol $119.37 \text{ g} = 0.495 \text{ mol}$ $CHCl_3$ d. molar mass of $C_2H_3Cl_3 = 133.39 \text{ g}$ 3.27 ...

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ACS statement on presidential proclamations limiting immigration. As one of the world's largest scientific societies, the American Chemical Society (ACS) expresses its concern with the June 22 presidential proclamation.

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