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Section 25.2 Nuclear Transformations 803 Number of Neutrons vs. Number of Protons for Stable Nuclei 20 40 60 80 100 Number of protons 120 20 0 40 60 80 100 Number of neutrons 120 140 Neutrons 1 Protons 1 (Ratio 1.0) Ratio 1.5 Ratio 1.3 Band of stability Ratio 1.2 Stable Nuclei 25.2 Nuclear Transformations Weather stripping and insula-

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25.2 Nuclear Transformations > 16 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. Nuclear Stability and Decay Nuclei that have an atomic number greater than 83 are radioactive. • These nuclei have both too many neutrons and too many protons to be stable. – Therefore, they undergo radioactive decay.

~~Chapter 25~~
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Number of Neutrons vs. Number of Protons for Stable Nuclei
20 40 60 80 100 Number of protons 120 20 0
40 60 80 100 Number of neutrons 120 140 Neutrons 1
Protons 1

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25.2 The first artificial transmutation reaction involved

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bombarding nitrogen gas with alpha particles. 19. 25.2 The elements in the periodic table with atomic numbers above 92, the atomic number of uranium, are called the transuranium elements. • All transuranium elements undergo transmutation.

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Nuclear Transformations. 2.1. RADIOACTIVITY.

Radioactivity, first discovered by Antonio Henri Becquerel (1852 to 1908) in 1896, is a phenomenon in which radiation is given off by the nuclei of the elements. ... we get the desired answer. EXAMPLE 3.

When will 5 mCi of ^{131}I ($T_{1/2} = 8.05$ days) and 2 mCi of ^{32}P ($T_{1/2} = 14.3$ days) have equal ...

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$1/2 \times 1/2 \times 1/2 \times 1/2 = 1/16$ of the original mass will remain. $1/16 \times 18.0 \text{ g} = 1.13 \text{ g}$ 23. $2.0\text{g} \times 1/2 \times 1/2 \times 1/2 = 0.25\text{g}$ or 3 half-lives Section Review 25.3 Part A Completion 1. fission 2. neutrons 3. fissionable atom

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Lecture 25.2- Nuclear Transformations

- Radon-222 is a radioactive isotope that is present naturally in the soil in some areas. It has a constant rate of decay. You will learn about decay rates of radioactive substances.

 25.2 4.

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Section 25.2 Radioactive Decay In your textbook, read about the changes that take place in an atomic nucleus when it decays. Circle the letter of the choice that best completes the statement. 1. The number of stable isotopes that exist compared to the number of unstable isotopes is a much less. b. much more. c. slightly more. d. about the same. 2.

~~Humble Independent School District / Homepage 25.2 Nuclear Transformations > 32 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. The stability of a nucleus depends on the ratio of neutrons to protons. Interpret Data • For~~

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elements of low atomic number (below about 20), this ratio is about 1. • Above atomic number 20, stable nuclei have more neutrons than protons.

~~Chapter 25~~

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